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# BaF<sub>2</sub>–contained tellurite glasses: Quantitative analysis and prediction of elastic properties and ultrasonic attenuation – Part II



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#### ARTICLE INFO ABSTRACT Keywords: The goal of the present paper is to complete our very recent work [J. Fluor. Chem. 210 (2018) 156-165] on the BaF2-doped tellurite glasses quantitative analysis and prediction of elastic properties, ultrasonic attenuation coefficient and glass transition Elastic properties temperature in binary BaF<sub>2</sub>-TeO<sub>2</sub> glass system over the composition range 7 mol $\% \leq$ BaF<sub>2</sub> $\geq$ 19 mol %. All the Ultrasonic attenuation above mentioned parameters were correlated with the ratio between packing density and mean atomic volume Rocherulle et al. model of the glass on the basis of Abd El-Moneim and Alfifi's approaches [2018]. In addition to this, the theoretical Abd El-Moneim and Alfifi's approaches elastic moduli and Poisson's ratio were evaluated on the basis of the Rocherulle et al. model [1989] and compared with the corresponding experimental ones. The results show that, the ratio between packing density and mean atomic volume can be considered as a new powerful tool capable of predicting changes in ultrasonic attenuation coefficient, elastic properties and glass transition temperature. Although both the theoretical and experimental elastic moduli display the same compositional trend; a slight discrepancy between the theoretical and experimental values of shear, longitudinal and Young's moduli is observed. This discrepancy has been attributed to the anomalous behavior between various elastic properties and calculated dissociation energy per unit volume. Finally, not only Rocherulle et al. model [1989] is valid for majority of the investigated glasses, but also the correlation ratios between theoretical and experimental elastic moduli are better than those reported

basis of Makishima-Mackenzie's theory [1973, 1975].

#### 1. Introduction

In recent years, tellurite-based glasses are studied intensively because of their various applications in electronics, nuclear and solar energy technologies, acousto-optic devices and optical fiber technology [1–6]. These glasses have several unique properties, such as good infrared transmittance, high refractive index and electrical semi-conductivity [7–11].

Fluoride-based heavy metal glasses gained much interest in the fields of fiber optics, quantum electronics and laser devices [12]. This type of glasses are known to have a wide variety of physical and optical properties such as high transparency from IR to UV region, low glass transition temperature, low linear and nonlinear refractive index, low phonon energies and their potential as host for active rare earth ions [13,14].

The choice of appropriate a glass for a particular purpose requires information about its mechanical / elastic properties. Mechanical properties, such as elastic moduli, Poisson's ratio and ultrasonic attenuation coefficient, are very informative about the structure of glasses, as they are affected by structural softening / compactness,

changes in geometrical configuration, dimensionality and cross-link density due to modifier doping. The ultrasonic methods have been already proved as an effective tool to study the fundamental mechanical / elastic properties of glasses. Wei [15] studied the elastic properties of silica glass doped with different concentration of fluorine. He observed a remarkable linear decrease in the measured density, ultrasonic velocities and elastic moduli with increasing fluorine concentration. More recently, Begum and Rajendran [16,17] have studied the effect of composition from the substitution of TeO<sub>2</sub> by the alkaline earth fluoride  ${\rm BaF}_2$  in  ${\rm BaF}_2\mbox{-}{\rm TeO}_2$  glasses or by the alkaline earth oxide BaO in BaO-TeO<sub>2</sub> glasses on the structure and properties of the tellurite network by means of XRD, SEM and DTA spectroscopes as well as the ultrasonic pulse-echo technique. The composition dependence of ultrasonic velocities and ultrasonic attenuation coefficient suggested that, both BaF<sub>2</sub> and BaO are incorporated the tellurite network as a glass modifier. This results in the conversion of TeO<sub>4</sub> trigonal bipyramids into TeO<sub>3+1</sub> polyhedral and then to TeO<sub>3</sub> trigonal pyramids through breaking of Te-eqOax-Te linkages and creation of non-bridging oxygen atoms (NBOs) [2,16,17]. In the first part of this work [18], we have reported a comprehensive study on the quantitative analysis and prediction of

previously when compared the same experimental elastic moduli with the theoretically calculated ones on the

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Experimental elastic modulus (GPa)

**Fig. 1.** Agreement between theoretical and experimental elastic moduli of binary  $BaF_2$ -TeO<sub>2</sub> glasses [18] on the basis of Makishima-Mackenzie's theory [20,21]. The solid line is the line of 1:1 correlation.

elastic properties and ultrasonic attenuation coefficient in binary  $BaF_2$ -TeO<sub>2</sub> glass system on the basis of the bond compression model [19], ring deformation model [19], Makishima-Mackenzie's theory [20,21] and Abd El-Moneim's model [22–24]. However, the following negative results were reported [18]:

- i The agreement between theoretical and experimental Young's, longitudinal and shears moduli is unsatisfactory for majority of the investigated glass samples as shown in Fig.1;
- ii The divergence between theoretical and experimental values increased with increasing BaF<sub>2</sub> content in the glass; and
- iii The behavior of all elastic properties with the calculated dissociation energy per unit volume of the glass is against Makishima-Mackenzie's theory [20,21].

However, Rocherulle et al. [25] have made an attempt to improve the agreement between the theoretical and experimental elastic moduli of multi-component oxide glasses; especially those have values of elastic moduli greater than 100 GPa. They made some modifications in the expression of the packing density and reported very good agreement between the theoretical and experimental elastic moduli. Very recently, Abd El-Moneim and Alfifi [26] have also been reported two new semiempirical formulas, which correlate both the ultrasonic attenuation coefficient and bulk modulus with the ratio between packing density and mean atomic volume of the glass. The same authors [26] have verified the applicability of their semi-empirical formulas for PbO-P<sub>2</sub>O<sub>5</sub>,  $V_2O_5$ -P<sub>2</sub>O<sub>5</sub>, CuO-P<sub>2</sub>O<sub>5</sub>, Cr<sub>2</sub>O<sub>3</sub>-doped Na<sub>2</sub>O-ZnO-P<sub>2</sub>O<sub>5</sub> and Ag<sub>2</sub>O-V<sub>2</sub>O<sub>5</sub>-TeO<sub>2</sub> glass systems.

As far the author knows, the applicability of Abd El-Moneim and Alfifi's semi-empirical formulas [26] and Rocherulle et al. model [25] has not been demonstrated for the  $BaF_2$ -doped tellurite glasses. Therefore, the present study had the following objectives:

- i To analyze and predict the composition dependence of ultrasonic attenuation coefficient, elastic properties and glass transition temperature in BaF<sub>2</sub>-TeO<sub>2</sub> glasses on the basis of Abd El-Moneim and Alfifi's approaches [26]. The correlations between all these parameters and the ratio between packing density and mean atomic volume were studied;
- ii To testify the applicability of Rocherulle et al. model [25] for the investigated  $BaF_2$ -TeO<sub>2</sub> glass samples. The theoretical values of elastic moduli and Poisson's ratio were calculated from the dissociation energy per unit volume and packing density and compared with the corresponding experimental values.

#### 2. Theory

The elastic properties and ultrasonic attenuation coefficient of a glass are affected by several different structural aspects. At the atomic scale, these parameters are closely related to the mean atomic volume and packing density. Nonetheless, in the past years, various simplest theoretical models were developed to predict the ultrasonic attenuation coefficient and to estimate various elastic properties of glasses based only on their chemical compositions. Among of these approaches, the presumably most frequently applied models were those proposed by Makishima and Mackenzie [20,21], Rocherulle et al. [25] and Abd El-Moneim and Alfifi [26].

### 2.1. Abd El-Moneim and Alfifi's approach [26] to ultrasonic attenuation coefficient and bulk modulus

In the first part of the present study [18], we expressed the ultrasonic attenuation coefficient of glassy materials on the basis of Rajendran et al. [27] suggestions as:

$$\alpha = a_1 \alpha_a + \alpha_c \tag{1}$$

where  $a_1$  is a constant, whereas  $\alpha_a$  and  $\alpha_c$  are the true absorption and coupling loss between the sample and ultrasonic transducer, respectively.

Very recently, Abd El-Moneim and Alfifi [26] reported the semiempirical formula (2), which correlates the experimentally measured ultrasonic attenuation coefficient at room temperature with the ratio between mean atomic volume and packing density of the glass;

#### Ultrasonic attenuation coefficient $\alpha = \chi / (V_t / \overline{V})^{\beta}$ (2)

Packing density 
$$V_t = \frac{1}{V_M} \sum_i x_i V_i$$
 (3)

Mean atomic volume  $\overline{V} = V_M/\psi$  (4)

Molar volume 
$$V_M = \sum_i x_i M_i / \rho$$
 (5)

Packing factor for a glass component *i* in the form  $A_a B_b$ 

$$V_i = 4\pi N_a \left( a R_A^3 + b R_B^3 \right) / 3 \tag{6}$$

where  $\chi$  and  $\beta$  are two constants, their values depend upon the glass composition and applied ultrasonic frequency, whereas  $N_a$ ,  $\rho$  and  $\psi$  are Avogadro's number, glass density and total number of atoms in the glass formula, respectively. On the other hand,  $x_i$  and  $M_i$  are the respective molar fraction and molecular weight of a glass component *i*, which have cation *A* and anion *B* with respective Pauling's ionic radii  $R_A$  and  $R_B$ . The advantage of Eq. (2) is that the ultrasonic attenuation coefficient is now expressed in terms of two compositional parameters, which are relatively easy to estimate from chemical composition of the glass compared with the calculation of the dissociation energy per unit volume ( $G_i$ ) and average first-order stretching force constant (*F*) of the glass, which were appeared a semi-empirical relationship reported previously by Abd El-Moneim [23] in the form;

$$\alpha = \frac{\chi'}{F} (F/10G_t V_t^2)^{m/n} \tag{7}$$

where  $\chi'$  is another constant, its values also depends upon the applied frequency and composition of the glass, whereas *m* and *n* are two positive powers, their values depends upon the glass-forming systems and their compositions.

Bulk modulus is defined as the ratio between isotropic pressure and relative volume change. On the basis of Abd El-Moneim and Alfifi's approaches [26], the bulk modulus of the glass can be expressed in terms of the ratio between mean atomic volume and packing density according to the following semi-empirical formula;

$$K = Q \ (V_t / \overline{V})^{\gamma}$$

where *Q* and  $\gamma$  are two new constants their values depend upon the type of the glass and its composition.

(8)

#### 2.2. Rocherulle et al. model [25] for elastic moduli and Poisson's ratio

Rocherulle et al. [25] tried to improve the agreement between theoretical and experimental elastic moduli of glasses; especially those have values of elastic moduli greater than 100 GPa. They made a modification in the expression of Makishima-Mackenzie's packing density [20,21] and reported the following semi-empirical equations, which relate the theoretical elastic moduli and Poisson's ratio of multicomponent oxide glasses with the packing density and dissociation energy per unit volume;

$$Young's modulus E_{th}^* = 8.36 C_t G_t$$
(9)

$$Bulk modulus K_{th}^* = 10.0 C_t^2 G_t$$
(10)

Shear modulus  $S_{th}^* = 3 K_{th}^* / (10.2 C_t - 1)$  (11)

Longitudinal modulus  $L_{th}^* = K_{th}^* + 4S_{th}^*/3$  (12)

Poisson's ratio  $\mu_{th}^* = 0.5 - (1/7.2 C_t)$  (13)

Packing density
$$C_t = \sum_i x_i V_i \rho_i / M_i$$
 (14)

Dissociation energy per unit volume

$$G_t = \sum_i x_i G_i \tag{15}$$

Dissociation energy per unit volume for a glass component *i* 

$$G_i = \rho_i U_i / M_i \tag{16}$$

where  $\rho_i$  and  $U_i$  are the respective density and molar dissociation energy of a glass component *i*. The semi-empirical Eqs. from (9) to (12) give the values of theoretical elastic moduli in GPa if the unit of  $G_t$  is kcal/cm<sup>3</sup>.

#### 3. Experimental

The various techniques for preparation of the investigated BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glass samples, measurement of density, ultrasonic velocities and attenuation coefficient and determination of the glass transition temperature were described in the earlier studies of Begum and Rajendran [16,17]. Al the ultrasonic measurements were carried out at room temperature and at 5 MHz frequency. The overall accuracy in density, ultrasonic velocity and ultrasonic attenuation coefficient measurements is  $\pm$  0.5 Kgm<sup>-3</sup>,  $\pm$  5% and  $\pm$  2%, respectively [16,17].

#### 4. Results and discussion

It is a well-known fact that, the network of pure tellurite glass is basically composed of TeO4 trigonal bipyramids, which are connected to each other through the covalent  $Te_{-eq}O_{ax}\text{-}Te$  linkages, where  $_{eq}O$ refers to oxygen in an equatorial plane and O<sub>ax</sub> refers to oxygen in axial position with respect to the Te atom [10]. In addition to this, each TeO<sub>4</sub> trigonal bipyramid has a lone pair of electrons in one of its equatorial sites [10]. The addition of a modifier to vitreous TeO<sub>2</sub> is expected to modify the network structure and changes all the physical and elastic properties of the glass. The type of modification depends upon the nature and concentration of the added modifier. The binary BaF2-TeO2 glasses studied in the present work are listed in Table 1 together with their compositions, ultrasonic properties and ultrasonic attenuation coefficient [16]. The literature available [16,18] on these glasses indicates that all the added molar fractions of the alkaline earth fluoride BaF2 are consumed in the transformation of TeO4 trigonal bipyramids into  $TeO_3$  trigonal pyramids through breaking of  $Te_{eq}O_{ax}$ -Te linkages

and creation of non-bridging oxygen atoms (NBOs) (two NBOs for each  $Ba^{2+}$  ion). Thus, the progressive increase in  $BaF_2$  and expense of  $TeO_2$  is expected to increase the number of NBOs, which consequently reduces the rigidity and compactness of  $BaF_2$ -TeO<sub>2</sub> glasses in comparison with the parent pure TeO<sub>2</sub> glass. However, in Ref. [18], we have calculated the number of network bonds per unit volume, average cross-link density and bond compression bulk modulus for the same present  $BaF_2$ -TeO<sub>2</sub> glasses from the bond compression model [19]. The results reveal that all these parameters show a gradual decrease with increasing  $BaF_2$  concentration.

The calculated values of packing density on the basis of Makishima-Mackenzie's theory [20,21], beside those of the mean atomic volume are given in Table 2 for BaF<sub>2</sub>-TeO<sub>2</sub> glass system under investigation. The necessary factors for calculating these parameters are given in Table 3. As clearly seen from this Table 2, the addition of the alkaline earth fluoride BaF<sub>2</sub> alters both the packing density and mean atomic volume. Increasing of BaF<sub>2</sub> from 7 to 19 mol% decreases the packing density from 0.5189 to 0.4896 and increases the mean atomic volume from 9.33 to 9.68 cm<sup>3</sup>/mol. The gradual decrease in the packing density with the substitution of TeO<sub>2</sub> by BaF<sub>2</sub> is related to the reduction in the rigidity and elastic stiffness of the glass due to the modifier role of BaF<sub>2</sub> as suggested by Makishima and Mackenzie [20,21].

The modifier role of the alkaline earth fluoride BaF<sub>2</sub> in the present BaF2-TeO2 glasses can also be understood from the variation of the mean atomic volume as follows. The increase in the mean atomic volume with the concurrent addition of  $\mbox{BaF}_2$  and reduction of  $\mbox{TeO}_2$  is expected to decrease the total number of atoms per unit volume of the glass ( $\psi N_a/V_M$ ), which consequently weakness and loose packing the structure in its resistance to compression as evidenced by the decrease of elastic moduli (Fig. 2) and glass transition temperature (Fig. 3) and increase of Poisson's ratio (Fig. 4). The increase in the ultrasonic attenuation coefficient supports the so-called open (loose packing and weak) structure of BaF<sub>2</sub>-TeO<sub>2</sub> glasses with the substitution of TeO<sub>2</sub> by BaF<sub>2</sub>. This discussion is in accordance with that reported in Ref. [18] when interpreting the compositional dependence of molar volume, packing density, fractal bond connectivity, average cross-link density, number of network bonds per unit volume, bond compression bulk modulus and mean atomic ring in the same glasses [18].

According to Abd El-Moneim and Alfifi's approaches [26], the ultrasonic attenuation coefficient and elastic moduli of glasses depend strongly upon the ratio between packing density and mean atomic volume. Thus, it is very important to testify these approaches [26] for the present  $BaF_2$ -TeO<sub>2</sub> glasses as follows.

### 4.1. Correlation between ultrasonic attenuation coefficient and the ratio between packing density and mean atomic volume

Ultrasonic attenuation coefficient describes the total reduction in sound intensity due to absorption of energy by the medium and the deflection of energy from the path of the beam by reflection, refraction and scattering [22]. The semi-empirical formula (2) suggested that, at a fixed frequency, a plot of ultrasonic attenuation coefficient versus the ratio between packing density and mean atomic volume  $(V_i/\overline{V})$  would yield an inverse proportionality. It is quite clear from Fig. 5 that, this is true for the present BaF<sub>2</sub>-TeO<sub>2</sub> glasses at 5 MHz frequency. The least-square linear regressions performed on  $\log(\alpha)$  and  $\log(V_i/\overline{V})$  yield the following semi-empirical relationship for ultrasonic attenuation coefficient

$$\alpha = 6.516 \times 10^{-6} / (V_t / \overline{V})^{3.933} \tag{17}$$

with correlation ratio of 98.4%,  $\chi = 6.516 \times 10^{-6}$  and  $\beta = 3.933$ . These results reveal that, the ratio between packing density and mean atomic volume play a dominate role in understanding and predicting changes in the ultrasonic attenuation coefficient of BaF<sub>2</sub>-TeO<sub>2</sub> glasses.

Previous studies [2,16-18] showed that, both BaF<sub>2</sub> and barium oxide BaO are incorporated the tellurite network as a glass modifier.

#### Table 1

	Compositions,	density, el	astic properties,	ultrasonic	attenuation	coefficient and	d glas	s transition	temperature	of bina	ry BaF₂∙	TeO <sub>2</sub>	glass system	ı [16	5].
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$T_g$	μ	E $K$ $L$ (GPa) (GPa)		L (GPa)	L S $\alpha_{5MHz}$ (GPa) (GPa) (dB/cm)		$\rho$ (g/cm <sup>3</sup> )	BaF <sub>2</sub>		
(K)		(014)	(014)			(ub) chirj	(8/ СШ )	(mol %)	(wt %)	
593	0.2654	48.09	34.20	59.5	19.0	0.57	5.7437	07	08	
591	0.2646	46.79	33.20	57.8	18.5	0.62	5.6698	09	10	
587	0.2693	45.44	32.90	56.7	17.9	0.75	5.6198	14	15	
583	0.2700	44.70	32.89	56.3	17.6	0.79	5.6028	17	18	
580	0.2733	44.57	32.83	56.1	17.5	0.81	5.6008	19	20	

Table 2

Compositional and theoretical elastic properties of binary BaF2-TeO2 glass system.

K <sub>bc</sub> [18] (GPa)	$\mu_{th}^*$	E <sub>th</sub> (GPa)	K <sub>th</sub> (GPa)	L <sub>th</sub> * (GPa)	S <sub>th</sub> (GPa)	$G_t$ [18] (kcal/cm <sup>3</sup> )	$C_t$	∇ (cm³/mol)	$V_t[18]$	BaF <sub>2</sub> (mol %)
73.25	0.2273	57.56	35.07	68.43	25.08	13.52	0.5093	9.33	0.5189	07
69.73	0.2254	57.88	35.01	68.61	25.26	13.69	0.5057	9.47	0.5095	09
59.19	0.2202	58.56	34.77	68.91	25.67	14.11	0.4964	9.60	0.4981	14
58.90	0.2171	58.99	34.64	69.13	25.93	14.37	0.4910	9.65	0.4925	17
56.34	0.2150	59.23	34.53	69.23	26.09	14.54	0.4873	9.68	0.4896	19

Table 3

Different parameters for glass components TeO2, BaO and BaF2 [18,28].

G <sub>i</sub> (kcal∕cm³)	Ci	V <sub>i</sub> (cm <sup>3</sup> /mole)	ρ <sub>i</sub> (g/cm <sup>3</sup> )	M <sub>i</sub> (g/mol)	Glass component
12.92	0.5222	14.70	5.670	159.6	TeO <sub>2</sub>
9.70	0.4887	13.10	5.720	153.33	BaO
12.92	0.3385	12.13	4.893	175.34	BaF <sub>2</sub>



Mean atomic volume (cm<sup>3</sup>/mol)

Fig. 2. Dependence of elastic moduli on mean atomic volume in binary  $BaF_{2}$ -TeO<sub>2</sub> glass system. The solid lines represent the least-square fitting of the data.

Thus, it is worth examining the validity of the same semi-empirical formula (2) for the binary BaO-TeO<sub>2</sub> glass system. The applied experimentally measured data of ultrasonic attenuation coefficient at room temperature and at 5 MHz frequency are listed in Table 4 for the investigated BaO-TeO<sub>2</sub> glass samples. The calculated values of packing density and mean atomic volume are listed in the same table. The applied values of  $V_i$  and  $M_i$  for BaO oxide are given in Table 3. Fig. 6 illustrates how the ultrasonic attenuation coefficient varies with the ratio  $(V_i/\nabla)$  in BaO-TeO<sub>2</sub> glasses at 5 MHz frequency. For comparison, the data of the present BaF<sub>2</sub>-TeO<sub>2</sub> glasses are included in the same figure. It



Mean atomic volume (cm<sup>3</sup>/mol)

Fig. 3. Dependence of glass transition temperature on mean atomic volume in binary  $BaF_2$ -TeO<sub>2</sub> glass system. The solid line represents the least-square fitting of the data.

is clear from this figure that, for both BaO-TeO<sub>2</sub> and BaF<sub>2</sub>-TeO<sub>2</sub> glasses, the ultrasonic attenuation coefficient shows a systematic decrease as the  $(V_t/\overline{V})$  ratio is increased. In addition to this, the fitted curve to the data of BaO-TeO<sub>2</sub> glasses is shifted towards lower values of ultrasonic attenuation coefficients. In case of BaO-TeO<sub>2</sub> glasses, the least-square linear regressions performed on log( $\alpha$ ) and log( $V_t/\overline{V}$ ) yield the following semi-empirical relationship for the ultrasonic attenuation coefficient at 5 MHz frequency;

$$\alpha = 19.186 \times 10^{-6} / (V_t / \overline{V})^{3.42}$$
(18)

with correlation ratio of 75%,  $\chi = 19.186 \times 10^{-6}$  and  $\beta = 3.42$ . These results proved that the semi-empirical formula (2) also satisfies the data of BaO-TeO<sub>2</sub> glasses very well. It noticeable that, although both BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses have a very close values of  $\beta$ , they have two different values of  $\chi$ . BaO-TeO<sub>2</sub> glasses have value of  $\chi$  greater than three times that of BaF<sub>2</sub>-TeO<sub>2</sub> glasses. These results are reasonable and agree with Abd El-Moneim and Alfifi's approaches [26]. However, the applicability of the semi-empirical formula (2) was verified very recently for PbO-P<sub>2</sub>O<sub>5</sub> and CuO-P<sub>2</sub>O<sub>5</sub> glasses at 10 MHz frequency, for V<sub>2</sub>O<sub>5</sub>-P<sub>2</sub>O<sub>5</sub> glasses at 8 MHz frequency, for Cr<sub>2</sub>O<sub>3</sub>-doped Na<sub>2</sub>O-ZnO-



Mean atomic volume (cm<sup>3</sup>/mol)

Fig. 4. Dependence of Poisson's ratio on mean atomic volume in binary  $BaF_{2}$ -TeO<sub>2</sub> glass system. The solid line represents the least-square fitting of the data.



Fig. 5. Relation between ultrasonic attenuation coefficient and the ratio between packing density and mean atomic volume glasses at 5 MHz in binary  $BaF_2$ -TeO<sub>2</sub> glass system. The solid line represents the least-square fitting of the data.

 $P_2O_5$  glasses at 5 MHz frequency and for Ag<sub>2</sub>O-V<sub>2</sub>O<sub>5</sub>-TeO<sub>2</sub> glasses at 4 MHz frequency [26]. The reported values of  $\chi$  and  $\beta$  for these glasses, along with those obtained in the present study for BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses are summarized in Table 5. From this table, we concluded that, for a given frequency, the values of  $\chi$  and  $\beta$  depend strongly upon the type of the glass and its composition. This supports Abd El-Moneim and Alfifi's approaches [26] and confirms the validity of the semi-empirical formula (2) for the phosphate- and tellurite-based glasses.

4.2. Correlating the elastic properties and glass transition temperature with the ratio between packing density and mean atomic volume

Fig. 7 shows how the experimental bulk modulus varies with the ratio  $(V_t/\overline{V})$  in the same binary BaF<sub>2</sub>-TeO<sub>2</sub> glass system under investigation. The forward proportionality between bulk modulus and the ratio  $(V_t/\overline{V})$  clarifies that the semi-empirical Eq. (8) satisfies the data of these glasses very well. The equation of the fitted curve can be represented by the semi-empirical formula;

$$K = 108.1 \ (V_t/\overline{V})^{0.4} \tag{19}$$

with correlation ratio of 85%, Q = 108.1 and  $\gamma = 0.4$ . These results suggest that the compositional dependence of bulk modulus in BaF<sub>2</sub>-TeO<sub>2</sub> glasses is predictable from Abd El-Moneim and Alfifi's approaches [26]. As expected, the obtained values of values of Q and  $\gamma$  for BaF<sub>2</sub>-TeO<sub>2</sub> glasses are different from those reported very recently for PbO-P<sub>2</sub>O<sub>5</sub>, V<sub>2</sub>O<sub>5</sub>-P<sub>2</sub>O<sub>5</sub> or the Cr<sub>2</sub>O<sub>3</sub>-doped Na<sub>2</sub>O-ZnO-P<sub>2</sub>O<sub>5</sub> glass systems as shown in Table 5. Fig. 8 illustrates how the bond compression bulk modulus ( $K_{bc}$ ) varies with the ratio between packing density and mean atomic volume in the same present BaF<sub>2</sub>-TeO<sub>2</sub> glasses. The data of  $K_{bc}$  were taken from our previous study on the same glasses [18]. The dependence of the bond compression bulk modulus on the ratio between packing density and mean atomic volume was fitted into a power semiempirical relation  $K_{bc} = 30.599 \times 10^3 (V_t/\overline{V})^{2.88}$  with correlation ration of 95.6%.

It is a well-known fact that, the glass transition temperature and Poisson's ratio reflect changes in the coordination number of the network forming atoms and destruction of the network structure brought about by the formation of NBOs in glasses. Thus, the composition dependence of these two parameters helps to reveal the loosely packed structure of the glass. Loosely packed glasses are expected to have lower glass transition temperature and higher Poisson's ratio compared with closely packed ones. In the light of this, the modifier role of  $BaF_2$  in the present  $BaF_2$ -TeO<sub>2</sub> glasses and that of BaO in BaO-TeO<sub>2</sub> glasses can be further support by correlating - the glass transition temperature and Poisson's ratio of these glasses with the ratio between packing density and mean atomic volume as follows:

i Fig. 9 shows the dependence of glass transition temperature  $(T_g)$  on the ratio  $(V_t/\overline{V})$  in BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses. The applied data of  $T_g$  were taken from Refs. [16,17]. The figure reveals clearly that, for both BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glass systems; there is a rapid increase in the glass transition temperature with increasing the ratio  $(V_t/\overline{V})$ . The relationship between glass transition temperature and  $(V_t/\overline{V})$  can be represented by the following semi-empirical equation;

$$T_g = P \left( V_t / \overline{V} \right)^{\delta} \tag{20}$$

where *P* and  $\delta$  are two constants, their values depend upon the type of the glass and its composition. The equations of the fitted curves to the data of BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses can be represented by the respective semi-empirical formulas (21) and (22),

Table 4

Composition	s, density	, ultrasonic attenuation of	coefficient, packin	g density	, mean atomic volume and s	glass transition tem	perature of binar	v BaO-TeO	glass sy	vstem
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1

μ[17]	$T_{g}[17]$	$\overline{V}$	Vt	$\alpha_{5MHz}[17]$	$\rho[17]$	BaO	
	(K)	(cm <sup>2</sup> /mol)		(db/cm)	(g/cm)	(mol %)	(wt %)
0.2630	599	9.69	0.51673	0.45	5.646	9.3	9
0.2653	591	9.78	0.51552	0.46	5.645	12.4	12
0.2694	585	9.93	0.51139	0.46	5.613	15.5	15
0.2740	580	10.07	0.50790	0.49	5.587	18.5	18
0.2764	576	10.15	0.50690	0.54	5.584	20.6	20
0.2779	579	10.22	0.50606	0.61	5.583	22.7	22



Ratio between packing density and mean atomic volume (mol/cm<sup>3</sup>)

Fig. 6. Relation between ultrasonic attenuation coefficient and the ratio between packing density and mean atomic volume glasses at 5 MHz in ( $\blacklozenge$ ) BaF<sub>2</sub>-TeO<sub>2</sub> glasses and ( $\Delta$ ) BaO-TeO<sub>2</sub>. The solid lines represent the least-square fitting of the data.

**Table 5**Values of the constants  $\chi$ , Q,  $\beta$  and  $\gamma$  for different glass systems.

Glass	Frequency	Using Eq. (2)	Using Eq. (8)		Ref.		
system	(MHZ)	X	β	Q γ			
BaF <sub>2</sub> -TeO <sub>2</sub>	5	$6.516\times10^{-6}$	3.933	108.1	0.40	Present study	
$BaO-TeO_2$	5	$19.186 \times 10^{-6}$	3.420	-	-	Present study	
CuO-P <sub>2</sub> O <sub>5</sub>	10	$80.00 \times 10^{-3}$	1.510	-	-	[26]	
PbO-P <sub>2</sub> O <sub>5</sub>	10	$1.11  imes 10^{-5}$	4.843	1272	1.44	[26]	
V <sub>2</sub> O <sub>5</sub> -P <sub>2</sub> O <sub>5</sub>	8	$3.98  imes 10^{-12}$	9.763	884	1.25	[26]	
Ag <sub>2</sub> O-V <sub>2</sub> O <sub>5</sub> - TeO <sub>2</sub>	4	$4.40 \times 10^{-9}$	7.229	-	-	[26]	
Cr <sub>2</sub> O <sub>3</sub> - doped Na <sub>2</sub> O- ZnO- P <sub>2</sub> O <sub>5</sub>	5	$5.20 \times 10^{-3}$	2.619	4752	4.77	[26]	



Ratio between packing density and mean atomic volume (mol/cm<sup>3</sup>)

Fig. 7. Relation between experimental bulk modulus and the ratio between packing density and mean atomic volume in binary  $BaF_2$ -TeO<sub>2</sub> glass system. The solid line represents the least-square fitting of the data.



Fig. 8. Relation between bond compression bulk modulus and the ratio between packing density and mean atomic volume in binary  $BaF_2$ -TeO<sub>2</sub> glass system. The solid line represents the least-square fitting of the data.



Ratio between packing density and mean atomic volume (mol/cm<sup>3</sup>)

Fig. 9. Relationship between glass transition temperature and the ratio between packing density and mean atomic volume in ( $\blacklozenge$ ) BaF<sub>2</sub>-TeO<sub>2</sub> glasses and ( $\Delta$ ) BaO-TeO<sub>2</sub> glasses. The solid lines represent the least-square fitting of the data.

$$T_g = 1134 (V_t/\overline{V})^{0.223} \tag{21}$$

with correlation ratio of 91.7%, P = 1134 and  $\delta = 0.223$ ;

$$T_g = 2359 (V_t / \overline{V})^{0.469} \tag{22}$$

with correlation ratio of 92.1%, P = 2359 and  $\delta = 0.469$ . According to these results, BaO-TeO<sub>2</sub> glasses have values of P and  $\delta$  greater than twice those of BaF<sub>2</sub>-TeO<sub>2</sub> glasses. The dependence of P and  $\delta$  the nature of the added modifier supports Abd El-Moneim and Alfifi's approaches [26].

ii Fig. 10 demonstrates the relationship between Poisson's ratio and  $(V_i/\overline{V})$  ratio in the same studied BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glass systems. The inverse proportionality between Poisson's ratio and  $(V_i/\overline{V})$  agrees with Abd El-Moneim and Alfifi's approaches [26]. These results suggest that Poisson's ratios of BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses are also predictable the ratio between packing density and mean atomic volume. The equations of the fitted curves can be represented by the semi-empirical formula

$$\mu = R / (V_t / \overline{V})^{\lambda}$$
<sup>(23)</sup>

where *R* and  $\lambda$  are two constants, their values depend upon the type



Ratio between packing density and mean atomic volume (mol/cm<sup>3</sup>)

Fig. 10. Relationship between Poisson's ratio and the ratio between packing density and mean atomic volume in ( $\blacklozenge$ ) BaF<sub>2</sub>-TeO<sub>2</sub> glasses and ( $\Delta$ ) BaO-TeO<sub>2</sub> glasses. The solid lines represent the least-square fitting of the data.

of the glass and its composition. The least-square linear regressions performed on  $\log(\mu)$  and  $\log(V_l/\overline{V})$  yield the respective semi-empirical relationships (24) and (25) for BaF<sub>2</sub>-TeO<sub>2</sub> and BaO-TeO<sub>2</sub> glasses;

$$\mu = 0.11 / (V_t / \overline{V})^{0.30} \tag{24}$$

with correlation ratio of 80.2%, R = 0.11 and  $\lambda = 0.30$ ;

$$\mu = 0.029 / (V_t / \overline{V})^{0.74} \tag{25}$$

with correlation ratio of 92.1%, R = 0.029 and  $\lambda = 0.74$ . In compared to BaF<sub>2</sub>-TeO<sub>2</sub> glasses, BaO-TeO<sub>2</sub> glasses have higher value of  $\lambda$  and smaller value of R. This confirms that Poisson's ratios of these glasses are predictable from the ratio between packing density and mean atomic volume. This discussion agrees very well with prediction of the same data of Poisson's ratio from the average cross-link density [18].

## 4.3. Agreement between theoretical and experimental elastic moduli on the basis of Rocherulle et al. model [25]

It is important to compare between the experimentally measured elastic moduli and Poisson's ratio of  $BaF_2$ -TeO<sub>2</sub> glasses with those theoretically calculated on the basis of Rocherulle et al. model [25]. The calculated values of packing density on the basis of Rocherulle et al. model [25], beside those of dissociation energy per unit volume, elastic moduli and Poisson's ratio are listed in Table 2 for all  $BaF_2$ -TeO<sub>2</sub> glass samples under investigation. The necessary factors for calculating both the packing density and dissociation energy per unit volume of the glass are given in Table 3 [18,28]. At first glance, the values of  $C_i$  follow the same compositional dependency which was seen before in the experimental data of elastic moduli [16,18]. This is in accordance with Rocherulle et al. [25] suggestions. The relationship between the theoretical and experimental elastic moduli of  $BaF_2$ -TeO<sub>2</sub> glasses is illustrated in Fig. 11. From the figure, we concluded that:

- i The theoretically calculated values of bulk modulus on the basis of Rocherulle et al. model [25] are in excellent agreement with the corresponding experimental values. The correlation ratio between the theoretical and experimental values ranges between 98% and 95%;
- ii There is a reasonable agreement between the theoretical and experimental values of longitudinal and Young's moduli. The



Experimental elastic modulus (GPa)

Fig. 11. Agreement between theoretical and experimental values of elastic moduli in binary  $BaF_2$ -TeO<sub>2</sub> glass system on the basis of Rocherulle et al. model [25]. The solid line is the line of 1:1 correlation.

correlation ratio ranges between 87% and 81% for longitudinal modulus and between ranges 84% and 75.2% for Young's modulus; iii The agreement between theoretical and experimental values of

shear modulus is unsatisfactory for all samples. The correlation ratio and between 76% and 67.1%.

However, in Ref. [18], we studied the agreement between the theoretical and experimental moduli elastic of the same studied  $BaF_2$ -TeO<sub>2</sub> glasses on the basis of Makishima-Mackenzie's model [20,21]. The correlation ratios between the theoretical and experimental values were found to be range between 94.2% and 93.4% for bulk modulus, between 86% and 81% for longitudinal modulus, between 82% and 75% for Young's modulus and between 75% and 67% for shears modulus. These results reveal that the model of Rocherulle et al. model [25] is more suitable than Makishima-Mackenzie's theory [20,21] for predicting the elastic moduli of the studied  $BaF_2$ -TeO<sub>2</sub> glasses. On the basis of Rocherulle et al. model [25], the agreement between theoretical and experimental values of Poisson's ratio was found to be ranges between 86% and 79%. These correlation ratios are very close to those reported previously for the same glasses [18] on the basis of Makishima-Mackenzie's model [20,21], which ranges between 87.6% and 79.1%.

One interesting observation in Fig. 11 is that, the theoretically calculated values of elastic moduli for all the investigated  $BaF_2$ -TeO<sub>2</sub> glass samples are greater than the corresponding experimental values. Moreover, the divergence between the theoretical and experimental values of elastic moduli and Poisson's ratio was found to be increases with increasing  $BaF_2$  concentration in the glass (see Tables 1 and 2). This can be explained as follows: Based on the modifier role of the alkaline earth fluoride  $BaF_2$  in the investigated  $BaF_2$ -TeO<sub>2</sub> glasses, the dissociation energy per unit volume of TeO<sub>2</sub> oxide was expressed as [18]

$$G_{TeO_2} = N_4 G_4 + (1 - N_4) G_3$$
(26)

where  $G_4$  and  $G_3$  are the respective dissociation energy per unit volume for TeO<sub>4</sub> and TeO<sub>3</sub> structural units, while  $N_4$  is the fraction of TeO<sub>4</sub> structural units in the glass. However, in the present calculation, we have ignored the presence of TeO<sub>3</sub> structural units and used the dissociation energy of 54 kJ/ cm<sup>3</sup> [28] for TO<sub>4</sub> groups, which has been determined empirically from Young's modulus of pure TeO<sub>2</sub> glass. This is because the concentration and dissociation energy per unit volume of the TeO<sub>3</sub> structural units are unknown so far. In addition to this, the experimental elastic moduli show opposite behavior with the calculated dissociation energy per unit volume of the glass as shown in Table 2.

This is against Rocherulle et al. model [25] and reflects in the divergence between the theoretical and experimental values of elastic moduli and Poisson's ratio.

From the above discussion, it can be concluded that the best correlation between experimental and theoretical elastic moduli and Poisson's ratio of  $BaF_2$ -TeO<sub>2</sub> glasses on the basis of Rocherulle et al. model [25] or Makishima-Mackenzie's theory [20,21] requires knowing the appropriate values of  $G_t$  and  $V_t$  for each investigated glass sample. Considering the effect of all the structural units, it is first necessary to identify all the types of bonds present in the prepared glass and then to ascribe appropriate values of  $G_i$  and  $V_i$  for each type of bonds by using neutron diffraction, Raman EXAFS and FTIR data. Thus, taking into account the effect of the basic structural units that present in the glass network as well as the uncertainty inherent in experimental measurements, we believe that both Makishima-Mackenzie's [20,21] and Rocherulle et al. models [25] are valid for the all the studied  $BaF_2$ -TeO<sub>2</sub> glasses and give good agreement between the calculated and observed values of elastic moduli and Poisson's ratio.

#### 5. Conclusions

A comprehensive study has been carried out on the prediction of ultrasonic attenuation coefficient, elastic properties and glass transition temperature in binary  $BaF_2$ -TeO<sub>2</sub> glass system. All the above mentioned parameters were correlated with the ratio between packing density and mean atomic volume of the glass on the basis of Abd El-Moneim and Alfifi's approaches [26]. The theoretical elastic moduli and Poisson's ratio have also been evaluated on the basis of Rocherulle et al. model [25] and compared with the corresponding experimental ones. The major conclusions can be summarized as follows:

- i The ratio between packing density and mean atomic volume can be considered as a new powerful tool capable of predicting the compositional dependence of ultrasonic attenuation coefficient, elastic properties and glass transition temperature in BaF<sub>2</sub>-TeO<sub>2</sub> glasses;
- ii Although both the theoretical and experimental elastic moduli display the same compositional trend, a discrepancy between the theoretical and experimental values of shear, longitudinal and Young's moduli is observed. This discrepancy has been interpreted in terms of the anomalous behavior between elastic moduli and calculated dissociation energy per unit volume of the glass; and
- iii In compared to Makishima-Mackenzie's theory [20,21], Rocherulle et al. model [25] appears to be more suitable for predicting the experimental data of elastic moduli.

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