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# Vacuum ultraviolet spectroscopic properties of $Pr^{3+}$ in $MYF_4$ (M = Li, Na, and K) and $LiLuF_4$

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## Abstract

LiYF<sub>4</sub>, NaYF<sub>4</sub>, KYF<sub>4</sub> and LiLuF<sub>4</sub> doped with Pr<sup>3+</sup> were synthesized by hydrothermal technique. Vacuum ultraviolet excitation and emission properties were investigated at 20 K. A new type of photon cascade emission was observed in  $LiYF_4$ :  $Pr^{3+}$  and  $LiLuF_4$ :  $Pr^{3+}$ , i.e. a first-step  $4f5d \rightarrow 4f^2$  emission followed by a second  $4f^2 \rightarrow 4f^2$  emission. In addition, the internal quantum efficiency of  $Pr^{3+}$  in NaYF<sub>4</sub> and  $KYF_4$  within visible range was proved to be larger than unity using the branching ratios data corrected by the emission spectra of  $YF_3$ :  $Pr^{3+}$ . © 2006 Elsevier B.V. All rights reserved.

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#### 1. Introduction

The photon cascade emission (PCE) phenomenon was first observed in YF<sub>3</sub> [1,2], and recently intense studies on the PCE materials have been triggered by the need for new vacuum ultraviolet (VUV) phosphors and scintillator materials. In order to observe PCE by Pr<sup>3+</sup>, it was required that the lowest energy 4f5d state was higher than the  ${}^{1}S_{0}$  state, otherwise PCE could not take place generally [3].

In this paper, we present another PCE process in which the first step is parity-allowed interconfigurational  $4f5d \rightarrow$ 4f<sup>2</sup> transition, and the second step is parity-forbidden intraconfigurational  $4f^2 \rightarrow 4f^2$  transition, even if the  ${}^{1}S_0$ state is higher than the lowest 4f5d state, the PCE still could be observed. Moreover, we get the internal quantum efficiency of Pr<sup>3+</sup>-doped NaYF<sub>4</sub> and KYF<sub>4</sub> in the visible range with branching ratios values corrected by YF<sub>3</sub>:Pr<sup>3+</sup> spectra.

## 2. Experimental

 $MYF_4$  (M = Li, Na, and K) and LiLuF<sub>4</sub> powders were synthesized by hydrothermal method. The starting materials were NH<sub>4</sub>HF<sub>2</sub> (A.R.), M<sub>2</sub>CO<sub>3</sub> (A.R.), Y<sub>2</sub>O<sub>3</sub> (99.99%) or Lu<sub>2</sub>O<sub>3</sub> (99.99%), Pr(NO<sub>3</sub>)<sub>3</sub>, and HF (A.R.). HF was added to modulate the pH value of starting reactant mixture to 3-4 and the details of hydrothermal method can be obtained from Ref. [4].

The products were checked by X-ray powder diffraction recorded on a Rigaku D/max-2400 powder diffractometer with CuKa radiation. The excitation and emission spectra were measured on the VUV Station of Beijing Synchrotron Laboratory (H7421-50 Photomultiplier) and National Synchrotron Radiation Laboratory (H5920-01 Photomultiplier) in University of Science and Technology of China. The excitation spectra were calibrated using sodium salicylate.

### 3. Results and discussion

The emission spectra of the investigated Pr<sup>3+</sup>-doped  $MYF_4$  (M = Li, Na, K) and LiLuF<sub>4</sub> are presented in

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Fig. 1. Emission spectra of  $Pr^{3+}$ -doped LiLuF<sub>4</sub> and MYF<sub>4</sub> (M = Li, Na, K) measured at 20 K.

Fig. 1. The emission spectra characteristics of  $Pr^{3+}$ -doped  $KYF_4$  and  $NaYF_4$  are similar to those of  $YF_3$ : $Pr^{3+}$  and typical for the systems where the  ${}^{1}S_{0}$  state of the Pr<sup>3+</sup> 4f<sup>2</sup> configuration lies below the lowest energy 4f5d state. Therefore, in the emission spectrum the bands at about 251, 272, 336, and 400 nm are observed, which are characteristic of the transitions from the  ${}^{1}S_{0}$  to the  ${}^{3}F_{4}$ ,  ${}^{1}G_{4}$ ,  ${}^{1}D_{2}$ , and  ${}^{1}I_{6}$  multiplets, respectively. In the visible range the emission line at 482 nm, corresponding to the  ${}^{3}P_{0} \rightarrow {}^{3}H_{4}$  transition, can be observed, indicating that in both NaYF4 and KYF4 the kind of PCE process consisting of two steps  ${}^{1}S_{0} \rightarrow {}^{3}P_{J} + {}^{1}I_{6}$ , and  ${}^{3}P_{0} \rightarrow {}^{3}H_{4}$  takes place. Furthermore, through comparison with the emission of  $Pr^{3+}$  in YF<sub>3</sub>, it can be concluded that the  ${}^{3}P_{0} \rightarrow {}^{3}H_{4}$ emission is more efficient in both hosts, especially for NaYF<sub>4</sub>: $Pr^{3+}$ .

The emission spectra of Pr<sup>3+</sup>-doped LiLuF<sub>4</sub>, LiYF<sub>4</sub> are dominating  $4f5d \rightarrow 4f^2$  emission. Accordingly, in the VUV range the band emission can be attributed to the  $4f5d \rightarrow {}^{3}H_{J} + {}^{3}F_{J}$  (about 250 nm),  $4f5d \rightarrow {}^{1}G_{4}$  (about 270 nm),  $4f5d \rightarrow {}^{1}D_{2}$  (about 346 nm),  $4f5d \rightarrow {}^{3}P_{I} + {}^{1}I_{6}$  (about 410 nm). In the visible range the stronger line at 483 nm due to the  ${}^{3}P_{0} \rightarrow {}^{3}H_{4}$  emission can also be observed under the  $4f^2 \rightarrow 4f5d$  excitation of  $Pr^{3+}$ . The appearance of this emission can be connected with the second stage of PCE. In this process there is the first  $4f5d \rightarrow {}^{3}P_{J} + {}^{1}I_{6}$ resulting in the wide-band emission around 410 nm. So we can conclude that the PCE in Pr<sup>3+</sup>-doped hosts arises not only from the  ${}^{1}S_{0}$  state but also from the lowest 4f5d state. Therefore, this quantum splitting effect has many important applications such as tunable UV laser and quantum cutter.

The emission spectra of  $Pr^{3+}$  in MYF<sub>4</sub> (M = Li, Na, K) are different. The fact is that in a first approximation the crystal-field splitting of 5d orbital decreases with the distance from a RE ion to its first neighbor anions, i.e., shorter is the distance, stronger is the splitting.



Fig. 2. Excitation spectra of  $Pr^{3+}$ -doped LiLuF<sub>4</sub> and MYF<sub>4</sub> (M = Li, Na, K) measured at 20 K.

In Fig. 2 we present the excitation spectra of  $Pr^{3+}$ -doped LiLuF<sub>4</sub> and MYF<sub>4</sub> (M = Li, Na, K) at 20 K and it can be concluded that  $Pr^{3+}$ -doped NaYF<sub>4</sub> and KYF<sub>4</sub> have similar high-energy configurations. At the same time there are analogous 4f5d states features in  $Pr^{3+}$ -doped LiLuF<sub>4</sub> and LiYF<sub>4</sub>. We have not observed the  ${}^{3}H_{4} \rightarrow {}^{1}S_{0}$  excitation line in all the hosts.

There is an important factor for the application of phosphors, i.e. quantum efficiency of visible range  $\eta_{total}$ , and from simple analysis which only considers the reduced matrix elements square  $\left\|U^{\lambda}\right\|^2$  and oscillator strengths we can conclude that the Judd–Ofelt parameters  $\Omega_2/\Omega_6$  and  $\Omega_4/\Omega_6$  determine the internal quantum efficiency of visible light  $\eta_{vis}$  [5]. Smaller these ratios, higher the branching ratio of  ${}^{1}S_{0} \rightarrow {}^{1}I_{6}$  and  $\eta_{vis}$ . The values of  $\Omega_{2}/\Omega_{6}$  and  $\Omega_{4}/\Omega_{6}$ can be obtained by fitting the experimentally corrected branching ratios. For difficulties in correcting emission spectra, we can get corrected value by comparing the emission spectra of YF<sub>3</sub>:Pr<sup>3+</sup> obtained from experimental measurement and Ref. [6]. The corrected branching ratios from the emission spectra for  $NaYF_4:Pr^{3+1}$ and  $KYF_4$ :  $Pr^{3+}$  have been given in Table 1. In the fitting procedure we find that the corrected branching ratios can be well reproduced by values of  $\Omega_2/\Omega_6$  and  $\Omega_4/\Omega_6$ . Hence we can roughly get  $\Omega_2/\Omega_6 = 0.015$  and  $\Omega_4/\Omega_6 = 0.08$  for NaYF<sub>4</sub>, at the same time  $\Omega_2/\Omega_6 = 0.013$  and  $\Omega_4/\Omega_6 = 0.07$ for KYF<sub>4</sub>. Then, we can conclude that the  $\eta_{vis}s$  for NaYF<sub>4</sub>:Pr<sup>3+</sup> and KYF<sub>4</sub>:Pr<sup>3+</sup> are larger than unity by comparing that of YF<sub>3</sub>:Pr<sup>3+</sup> [6] ( $\eta_{vis} = 157\%$  obtained from  $\Omega_2/\Omega_6 = 0.013$  and  $\Omega_4/\Omega_6 = 0.07$ ). Therefore,  $Pr^{3+}$ doped NaYF<sub>4</sub> and KYF<sub>4</sub> are also suitable for PCE materials.

## 4. Conclusions

LiLuF<sub>4</sub>: $Pr^{3+}$  and MYF<sub>4</sub>: $Pr^{3+}$  (M = Li, Na, K) were synthesized by hydrothermal method. The VUV excitation

Emission ${}^{1}S_{0} \rightarrow$	YF <sub>3</sub> :Pr <sup>3+</sup>			NaYF <sub>4</sub> :Pr <sup>3+</sup>		KYF <sub>4</sub> :Pr <sup>3+</sup>	
	$\beta_{\rm YF_3}$ [6]	$\beta_{exp,YF_3}$	$C = \beta_{\rm YF_3} / \beta_{\rm exp, YF_3}$	$\beta_{\rm exp,NaYF_4}$	$\beta_{\rm corre,NaYF_4}$	$\beta_{\exp,KYF_4}$	$\beta_{\rm corre,KYF_4}$
<sup>1</sup> I <sub>6</sub>	0.789	0.709	1.11	0.676	0.776	0.763	0.847
$^{1}D_{2}$	0.015	0.045	0.333	0.052	0.018	0.051	0.017
$^{1}G_{4}$	0.128	0.167	0.766	0.186	0.147	0.101	0.077
${}^{3}F_{4}$	0.053	0.079	0.671	0.086	0.06	0.084	0.056

Table 1 Transition branching ratios and correction factors for the determination of  $\eta_{vis}$ 

 $\beta_{\rm YF_3}$ : branching ratios data (corrected) for YF<sub>3</sub>:Pr<sup>3+</sup> [6],  $\beta_{\rm exp,YF_3}$ :measured (uncorrected) branching ratios for YF<sub>3</sub>:Pr<sup>3+</sup> using our setup, *C*: correction factors corresponding to our experimental setup,  $\beta_{\rm exp,NaYF_4}$ : measured (uncorrected) branching ratios for NaYF<sub>4</sub>:Pr<sup>3+</sup>,  $\beta_{\rm corre,NaYF_4}$ : corrected branching ratios by use of *C* for NaYF<sub>4</sub>:Pr<sup>3+</sup>.  $\beta_{\rm exp,KYF_4}$ : measured (uncorrected) branching ratios for KYF<sub>4</sub>:Pr<sup>3+</sup>,  $\beta_{\rm corre,KYF_4}$ : corrected branching ratios using *C* for KYF<sub>4</sub>:Pr<sup>3+</sup>.

and emission spectra indicated that there was another PCE process in the LiLuF<sub>4</sub>:Pr<sup>3+</sup> and LiYF<sub>4</sub>:Pr<sup>3+</sup>, i.e. the first step was  $4f5d \rightarrow 4f^2$  emission, and the second step was the  $4f^2 \rightarrow 4f^2$  emission. Furthermore, the internal quantum efficiency of Pr<sup>3+</sup>-doped NaYF<sub>4</sub> and KYF<sub>4</sub> emission in the visible range was proved to be larger than unity. Then it was concluded that NaYF<sub>4</sub>:Pr<sup>3+</sup> and KYF<sub>4</sub>:Pr<sup>3+</sup> were also the potential efficient PCE materials.

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