



Effects of LiF on sintering characteristics and dielectric properties of low-loss SrCuSi₄O₁₀ ceramics for LTCC applications

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HIGHLIGHTS

- The sintering temperature of SrCuSi₄O₁₀ ceramics reduced to 900 °C from 1100 °C.
- The main peak exhibited a slight shift toward the higher 2θ degrees.
- The τ_e can be effectively modified to near zero by changing the TiO₂ content.
- SrCuSi₄O₁₀-1.0 wt% LiF-21 wt% TiO₂ ceramics exhibited good dielectric properties.

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ABSTRACT

The SrCuSi₄O_{10-x} wt% LiF (x = 0.5, 1.0, 1.5, 2.0) ceramics were prepared by the conventional solid-state route. The effects of LiF on the sintering behavior, microstructure, phase evolution and dielectric properties of the SrCuSi₄O₁₀ ceramics were systematically investigated. And the sintering temperature of SrCuSi₄O₁₀ ceramics has been reduced to 900 °C from 1100 °C with a certain amount of LiF. The SrCuSi₄O₁₀ ceramics with 1.0 wt% LiF addition sintered at 900 °C for 6 h exhibited excellent dielectric properties of ε_r ~ 5.88, tanδ ~ 1.6 × 10⁻³, τ_e ~ 119.90 ppm/°C. However, the relatively large positive τ_e was unfavorable to practical applications. Hence, the TiO₂ that owned a considerable negative τ_e was introduced to obtain a desired τ_e value. The addition of 21 wt % TiO₂ effectively improved the temperature stability of ceramics and τ_e was adjusted to near zero. The prepared SrCuSi₄O₁₀-1.0 wt% LiF-21 wt% TiO₂ ceramics sintered at 900 °C for 6 h showed fairly good dielectric properties of ε_r ~ 6.73, tanδ ~ 3.7 × 10⁻³, τ_e ~ 1.80 ppm/°C.

1. Introduction

With the development of wireless communication systems, microwave dielectric materials have attracted great interests for advantages on passive components, such as resonators, antennas and filters [1]. In order to satisfy the demands for miniaturization, packaging and integration of various microwave devices, the low temperature co-fired ceramic (LTCC) technology has attracted considerable attentions owing to the ability to combine the multi-layer ceramics and conductors for various modules and substrates [2–5]. For practical applications, the LTCC substrate materials should possess low dielectric loss (tanδ), low dielectric constant (ε_r) to reduce signal delay, and near-zero temperature coefficient of dielectric constant (τ_e) for stability [6–8]. Silicates, a candidate of substrate materials, own low ε_r due to the silica-oxygen tetrahedra composed of half covalent bonds [9,10]. Wesselsite (SrCuSi₄O₁₀), which was first systematically reported by Manu et al.,

showed a low ε_r ~ 5.05 and tanδ ~ 9.6 × 10⁻⁴ at 1 MHz when sintered at 1100 °C for 6 h [11]. However, the required high sintering temperature of SrCuSi₄O₁₀ ceramics restricted the practical applications in the LTCC field. The sintering temperature of LTCCs should be lower than the melting point of the common electrode materials, such as Ag (961 °C) and Cu (1083 °C) [12,13]. Therefore, it is significant to lower the sintering temperature of ceramics to meet the requirements of LTCC applications.

Recently, in order to lower the sintering temperature of ceramics, many strategies have been proposed. The approach of adding low melting point materials to ceramics is widely used to lower the densification temperature of dielectric ceramics [14–16]. It is well known that lithium fluorite (LiF) is an effective sintering aid for various materials. Li⁺ is an easily diffused ion at low temperature, which makes it be a very efficient sintering additive for many materials [17–20]. What's more, in order to attain the temperature compensation of

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dielectric constant, the materials with opposite τ_e are used to obtain a desired τ_e value. In our previous study, the τ_e of SrCuSi₄O₁₀ ceramics is a relatively large and positive value [21]. Therefore, we should introduce some material that owns a big and negative τ_e value, such as TiO₂, to tune the τ_e to near zero.

In this work, LiF was used as sintering aid and TiO₂ was employed as τ_e compensator. We took two steps to obtain the low-temperature co-fired ceramics with excellent dielectric properties. Firstly, a small amount of LiF was considered to lower the densification temperature of SrCuSi₄O₁₀ ceramics. Then, the SrCuSi₄O₁₀-LiF ceramics were mixed with TiO₂ to tune the τ_e . The effects of LiF on the sintering characteristics, microstructures and dielectric properties of SrCuSi₄O₁₀ ceramics with addition of TiO₂ were investigated systematically.

2. Experimental procedure

The SrCuSi₄O₁₀ ceramics were synthesized by the conventional solid-state reaction method using high purity powders of SrCO₃ (99.0%), CuO (99.0%) and SiO₂ (99.0%). The starting materials were ball-milled with distilled water for 6 h according to the stoichiometric compositions of SrCuSi₄O₁₀. The mixtures were dried, crushed and sieved with a 40 mesh screen. Then the sieved samples were calcined at 975 °C for 4 h in air. Afterwards, SrCuSi₄O₁₀-*x* wt% LiF (*x* = 0.5, 1.0, 1.5, 2.0) and SrCuSi₄O₁₀-1.0 wt% LiF-*y* wt% TiO₂ (*y* = 15, 18, 21, 24) mixtures were prepared by pure SrCuSi₄O₁₀, LiF and TiO₂ powders and re-milled for 6 h. After dried, paraffin was added to the powders as a binder. Pellets of 15 mm in diameter and 6–7 mm in thickness were prepared under a pressure of 4 MPa by uniaxial pressing. Then, the obtained pellets were sintered at 800–950 °C for 6 h in air with the heating rate of 3 °C/min.

The bulk densities of the sintered ceramics were measured by the Archimedes method. The crystalline phases were identified by X-ray diffraction (XRD) (Rigaku D/max 2550 PC, Tokyo, Japan) with Cu K α radiation (*V* = 200 kV, *I* = 40 mA) over a 2 θ angle from 10° to 70°. The microstructures of the sintered samples were observed and analyzed by a scanning electron microscopy (SEM) (MERLIN Compact, Germany). The microwave dielectric properties of sintered samples were measured by a high precision LCR meter (Agilent E4981A, USA) at a frequency of 1 MHz on silver-plated pellets. The τ_e was also measured by the LCR meter (Agilent E4981A, USA) at a frequency of 1 MHz with a thermostat in the temperature range from 25 °C to 85 °C and was calculated by the following formula:

$$\tau_e = \frac{\varepsilon_{85} - \varepsilon_{25}}{\varepsilon_{25}(85 - 25)} \times 10^6 (\text{ppm}/^\circ\text{C})$$

Where ε_{25} and ε_{85} were the dielectric constants at 25 °C and 85 °C, respectively.

3. Results and discussion

3.1. SrCuSi₄O₁₀-LiF system

Fig. 1 showed the X-ray diffraction patterns of SrCuSi₄O₁₀ ceramics with 0.5, 1.0 and 1.5 wt% LiF sintered at 900 °C and 2.0 wt% LiF sintered at 875 °C for 6 h in air respectively. The peaks were indexed by the standard JCPDS file for SrCuSi₄O₁₀ (PDF #49-1813) and no additional peaks were observed. That indicated that there was no chemical reaction between SrCuSi₄O₁₀ and LiF. The main peak (202) shift was enlarged separately and plotted in Fig. 1. It was interesting to note that the main peak (202) exhibited a slight shift toward the higher 2 θ degrees with increasing LiF addition when sintered at 900 °C shown in Fig. 1. As was well known, the shift of the peak position was associated with the variation of cell volume. Thus, the unit cell volume decreased due to the substitution of the F⁻ ion in the O²⁻ sites considering their relative ionic sizes (radius F⁻ = 1.33 Å, radius O²⁻ = 1.40 Å) [22]. When adding 2.0 wt% LiF to the samples sintered at 875 °C, the main

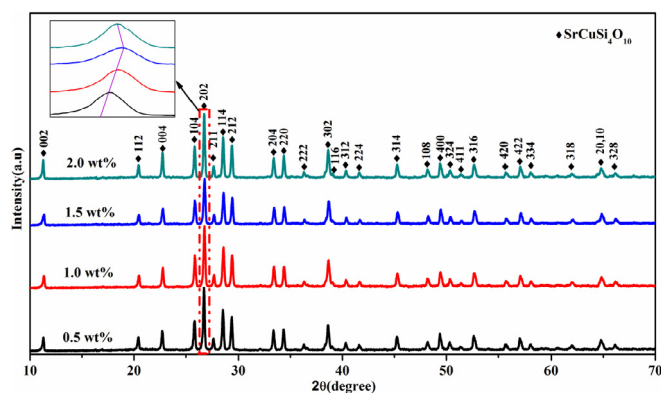


Fig. 1. The XRD patterns of the SrCuSi₄O₁₀ ceramics sintered at 900 °C for 6 h with (a) 0.5 wt%, (b) 1.0 wt%, (c) 1.5 wt% and sintered at 875 °C with (d) 2.0 wt% LiF.

peak (202) slightly shifted to a lower degree, one possible assumption was that the liquid phase during sintering process inhibited the entrance or occupation of F⁻ in the SrCuSi₄O₁₀ lattice.

Typical SEM micrographs of the surfaces of SrCuSi₄O₁₀-*x* wt% LiF ceramics sintered at different temperatures were illustrated in Fig. 2. For the samples sintered at 900 °C, the pores decreased with the increasing *x*, as shown in Fig. 2 (a)–(c). When 0.5 wt% LiF was added, the ceramic samples sintered at 900 °C were highly porous, which indicated that the amount of the liquid phase was insufficient for sintering of the SrCuSi₄O₁₀ ceramics. In contrast to that, in the samples added with 1.0–1.5 wt% LiF sintered at low temperature, no obvious pores could be detected, which indicated the addition of LiF in improving the sintering behavior was effective. However, for the samples containing 2.0 wt% LiF, a higher porosity of sample surface was observed, which led to a decrease in microwave dielectric properties. Therefore, it was demonstrated that while a moderate amount of LiF improved the sinterability at low temperature, an excessive addition of sintering aid would remain in grain boundary.

Fig. 3 illustrated the bulk densities of SrCuSi₄O₁₀ ceramics with LiF addition as a function of sintering temperature. The concentration of the LiF strongly influenced the bulk density of the samples and the densification temperature was sharply lowered with the increasing LiF content. It could be observed that the bulk densities of SrCuSi₄O₁₀ ceramics increased with the increment of sintering temperature. And the density of the samples ascended to a maximum value of 3.06 g/cm³,

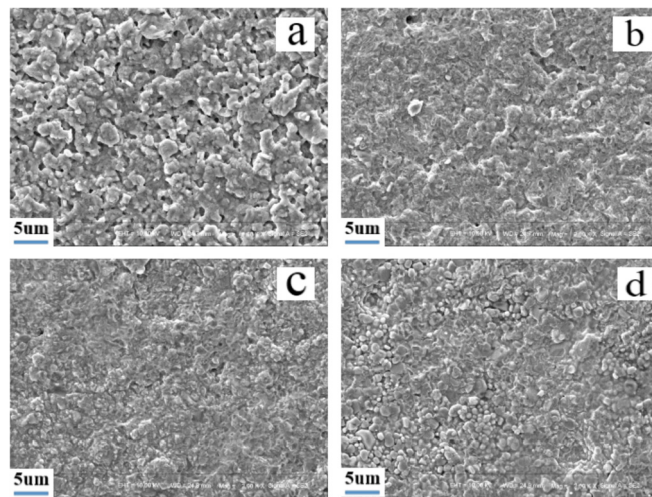


Fig. 2. The SEM micrographs of SrCuSi₄O₁₀-*x* wt% LiF ceramics sintered at 900 °C for 6 h: (a) *x* = 0.5 (b) *x* = 1.0 (c) *x* = 1.5, and sintered at 875 °C: (d) *x* = 2.0.

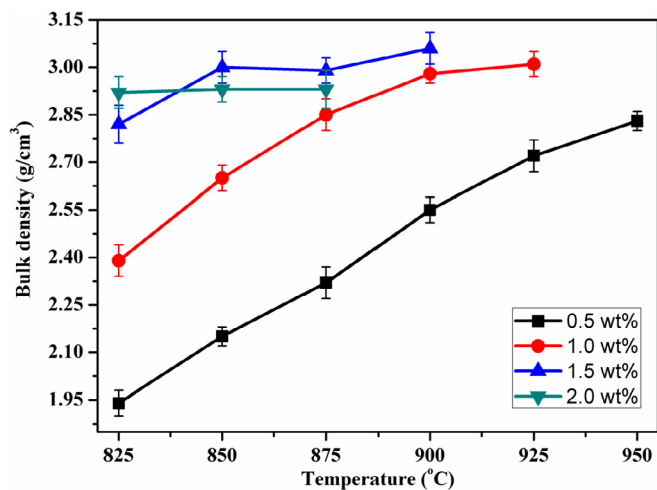


Fig. 3. The bulk density of SrCuSi₄O₁₀ ceramics with x wt% LiF as a function of sintering temperature.

when the SrCuSi₄O₁₀ ceramics with 1.5 wt% LiF addition were sintered at 900 °C. Moreover, the bulk density of the samples increased sharply with the sintering temperature when low level LiF (0.5 wt%) was added, which indicated that the 0.5 wt% LiF was not enough to improve the densification of SrCuSi₄O₁₀ ceramics effectively at the relatively low sintering temperatures. For the specimens with $x = 2.0$, with the increase of temperature, the changes of bulk density were not obvious. It was well known that a large extent of liquid phase was not favorable for mass transport and LiF had a relatively lower theoretical density of 2.635 g/cm³ compared with the SrCuSi₄O₁₀ ceramics. Therefore, for the samples with higher content of LiF, the bulk density was not significantly improved. All the results demonstrated that the LiF additive was an effective sintering aid for SrCuSi₄O₁₀ ceramics.

Fig. 4 showed the dielectric constant of SrCuSi₄O₁₀ ceramics with different amount of LiF addition as a function of sintering temperature. Obviously, the relationship between ϵ_r value and sintering temperature showed the same trend as the relationship between bulk density and sintering temperature. It was known that the ϵ_r usually depended on the bulk density, second phase and porosity [23]. In this work, it was obvious that the ϵ_r values were mainly dependent on the bulk density and pores due to the obtained single phase as shown in Fig. 1. Thus, the SrCuSi₄O₁₀-0.5 wt% LiF ceramics with a low bulk density had a lower dielectric constant than others.

The dielectric loss of the SrCuSi₄O₁₀ ceramics at 1 MHz with LiF

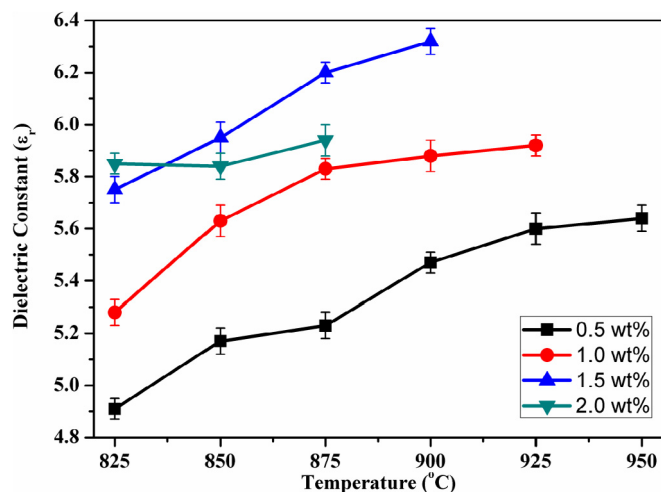


Fig. 4. The dielectric constant of SrCuSi₄O₁₀ ceramics with x wt% LiF as a function of sintering temperature.

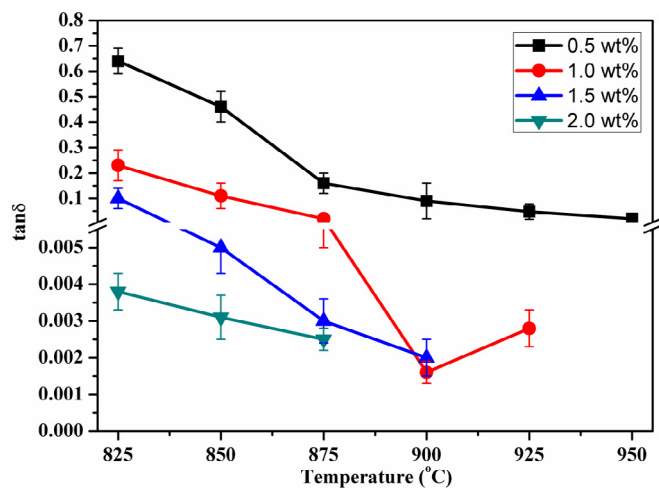


Fig. 5. The tan δ values of SrCuSi₄O₁₀ ceramics with x wt% LiF as a function of sintering temperature.

addition as a function of sintering temperature was shown in Fig. 5. It was well known that dielectric loss was mainly caused not only by the lattice vibration modes but also by porosity, second phase, lattice defect and grain boundary or grain morphology [23–26]. LiF-added samples, the variation of tan δ with sintering temperatures was similar to that of bulk density, suggested that the density was the dominating factor to control tan δ in SrCuSi₄O₁₀ ceramics. For slight amount of LiF-added samples ($x = 0.5$), it could be observed that the tan δ values decreased with the increasing sintering temperature. By contrast, the tan δ value of SrCuSi₄O₁₀ ceramics with 1.0 wt% LiF decreased with the increasing sintering temperature then increased after reaching their respective minimum value. The decrease in the tan δ value could be solely attributed to an increase in the density, whereas its increase at higher temperature was due to the excessive liquid phase by LiF addition at the grain boundaries. This illustrated that the amount of additives required less and less as the temperature increased in order to achieve the same density and excellent dielectric properties. And the results of SrCuSi₄O₁₀ ceramics with 1.5 and 2.0 wt% LiF also proved this. For SrCuSi₄O₁₀ ceramics with 1.5 wt% LiF, the samples melted in the container when the sintering temperature exceeded 900 °C. The maximum sintering temperature of SrCuSi₄O₁₀ ceramics with 2.0 wt% LiF was 875 °C. Typically, the optimum tan $\delta \sim 1.6 \times 10^{-3}$ at 1 MHz was obtained in SrCuSi₄O₁₀-1.0 wt% LiF ceramics sintered at 900 °C.

Fig. 6 indicated the temperature coefficients of dielectric constant values of SrCuSi₄O₁₀ ceramics sintered at 875 °C and 900 °C as a function of LiF content. The τ_e values basically increased nonlinearly to more positive values as x increased and shifted from 121.08 to 132.96 ppm/°C with LiF content increasing from 0.5 to 2.0 wt% sintered at 875 °C. And the changes of τ_e values were not obvious between different temperatures, which indicated that the τ_e values were not sensitive to the sintering temperature. It was well known that the temperature coefficient predominantly depended on crystal structure and lattice parameter of the materials [27]. Therefore, it could be reasonably believed that the variation of τ_e values should be closely related to the substitution of the F⁻ ion in the O²⁻ site, which resulted in the unit cell volume decreased.

3.2. SrCuSi₄O₁₀-LiF-TiO₂ system

In order to adjust the τ_e close to zero, TiO₂ was added into SrCuSi₄O₁₀ -1.0 wt% LiF samples. Fig. 7 showed the XRD patterns of SrCuSi₄O₁₀ -1.0 wt% LiF ceramic modified with 15–24 wt% TiO₂ sintered at 900 °C in air for 6 h. Besides the main phase SrCuSi₄O₁₀, an amount of TiO₂ phase were detected when TiO₂ was added. And the variation of the intensity of secondary phases was not obvious with

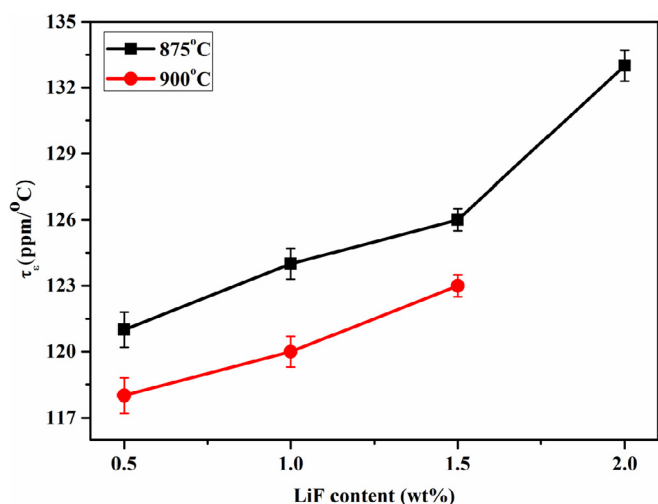


Fig. 6. The τ_e values of SrCuSi₄O₁₀ ceramics with x wt% LiF sintered at 900 °C and 875 °C for 6 h.

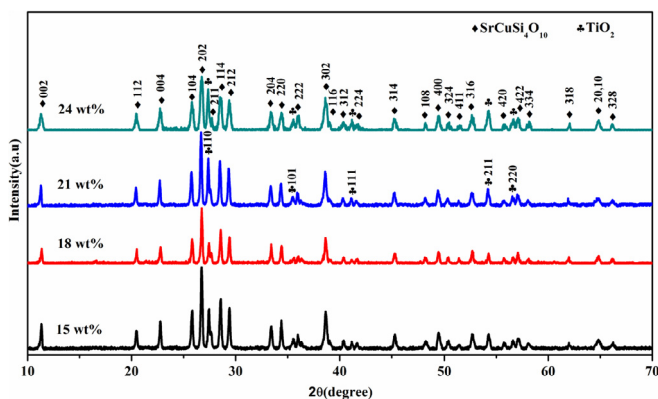


Fig. 7. The XRD patterns of the SrCuSi₄O₁₀-1.0 wt% LiF ceramics sintered at 900 °C for 6 h with (a) 15 wt%, (b) 18 wt%, (c) 21 wt% and (d) 24 wt% TiO₂.

increasing the TiO₂ content.

The bulk density, dielectric constant, and dielectric loss of SrCuSi₄O₁₀ ceramics as a function of the TiO₂ content were shown in Fig. 8. It could be observed that the bulk densities increased, as the TiO₂ content increased from 15 to 24 wt%. And the bulk density increased from 3.05 to 3.21 g/cm³ with the changing of TiO₂ content because TiO₂ possessed a higher bulk density (4.236 g/cm³). The ϵ_r value revealed rather similar tendency with the bulk density, since the higher bulk density meant lower porosity. Besides, it was well known that TiO₂ had a higher dielectric constant ($\epsilon_r \sim 105$) [28]. Therefore, it could also explain why the ϵ_r value increased from 6.42 to 6.85. In addition, a sustained upward trend in the $\tan\delta$ value increasing from 2.7×10^{-3} to 5.4×10^{-3} was observed, as the TiO₂ content increased.

Fig. 9 showed the temperature coefficients of the dielectric constant of SrCuSi₄O₁₀ ceramics sintered at 900 °C as a function of the TiO₂ content. When the content of TiO₂ was increased, the τ_e value was observed to decrease from 38.60 to -15.40 ppm/°C. Furthermore, a near zero τ_e value (1.80 ppm/°C) in this system was obtained when the SrCuSi₄O₁₀-1.0 wt% LiF samples added with 21 wt% TiO₂. As we known, the τ_e value of the ceramics was greatly related to the composition of the ceramics. According to the study above, the SrCuSi₄O₁₀-1.0 wt% LiF ceramics with y wt% TiO₂ possessed positive τ_e values and negative τ_e values. Therefore, the τ_e value in this system could be effectively adjusted from positive to negative by changing the content of TiO₂.

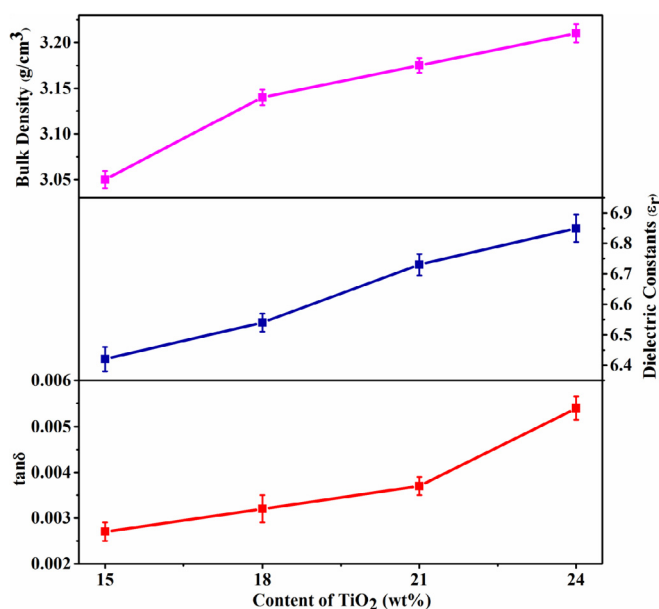


Fig. 8. Bulk density, dielectric constant and $\tan\delta$ of SrCuSi₄O₁₀-1.0 wt% LiF ceramics with y wt% TiO₂ sintered at 900 °C for 6 h.

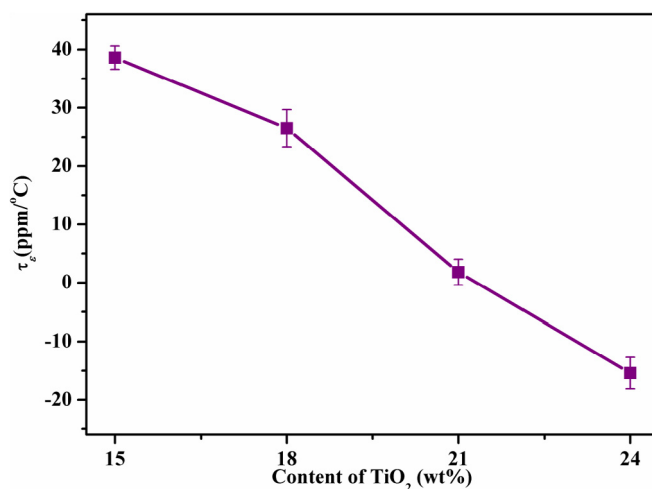


Fig. 9. The τ_e values of SrCuSi₄O₁₀-1.0 wt% LiF ceramics with y wt% TiO₂ sintered at 900 °C for 6 h.

4. Conclusion

In this study, the effects of LiF addition on the sintering behaviors and dielectric properties of SrCuSi₄O₁₀ ceramic sintered at low temperature were investigated. And the sintering temperature of SrCuSi₄O₁₀ ceramics reduced to 900 °C from 1100 °C added with 1.0 wt% LiF. The main peak (202) exhibited a slight shift toward the higher 2θ degrees with increasing LiF addition due to the substitution of the F⁻ ion in the O²⁻ sites. The behaviors of the bulk density of SrCuSi₄O₁₀ ceramics were associated with the sintering temperature and the amount of LiF. And the ϵ_r was dependent on the bulk density. The $\tan\delta$ values were mainly affected by the extrinsic losses (densification and grain boundaries). Typically, excellent dielectric properties with $\epsilon_r \sim 5.88$, $\tan\delta \sim 1.6 \times 10^{-3}$ and $\tau_e \sim 119.90$ ppm/°C were obtained for SrCuSi₄O₁₀-1.0 wt% LiF ceramics sintered at 900 °C. On this basis, τ_e could be effectively modified to near zero by changing the TiO₂ content in SrCuSi₄O₁₀-1.0 wt% LiF ceramics and a stable two-phase system SrCuSi₄O₁₀-TiO₂ was formed. When the content of TiO₂ was increased from 15 to 24 wt%, the τ_e decreased from 38.60 to -15.40 ppm/°C.

Addition of 21 wt% TiO₂ effectively improved the temperature stability of ceramics and τ_e was adjusted to near zero. SrCuSi₄O₁₀-1.0 wt% LiF-21 wt% TiO₂ ceramics exhibited good dielectric properties of $\epsilon_r \sim 6.73$, $\tan\delta \sim 3.7 \times 10^{-3}$ and $\tau_e \sim 1.80$ ppm/°C when sintered at 900 °C for 6 h.

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