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Radiation Measurements

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Effects of Na and K co-doping on growth and scintillation properties of Eu:SrI₂ crystals



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HIGHLIGHTS

- Na or K co-doped Eu:SrI₂crystals were grown by the modified μ-PD method.
- The milky parts were generated in the Na5.0% and K5.0%, Eu: SrI₂ crystals.
- The light yield of Eu:SrI₂was decreased by the Na or K co-doping.
- The decay times of Eu:Srl₂were almost constant by the Na or K co-doping.
- The non-proportionalitywas improved in the low energy region by the K co-doping.

ARTICLE INFO

Article history: Received 25 October 2015 Received in revised form 20 January 2016 Accepted 21 January 2016 Available online 25 January 2016

Keywords: Eu:Srl₂ Scintillator Single crystal Alkali metal Light yield Non-proportionality

ABSTRACT

We grew Na and K co-doped Eu:Srl₂ [Na,Eu:Srl₂ and K,Eu:Srl₂] crystals by a modified micro-pulling-down method to reveal the co-doping effects on the crystal growth and scintillation properties. The non-codoped, Na0.5%, Na1.0%, K0.5% and K1.0%,Eu:Srl₂ crystals indicated high transparency while the milky parts were generated in the Na5.0% and K5.0%,Eu:Srl₂ crystals. The light yields of Na,Eu:Srl₂ and K,Eu:Srl₂ crystals under γ -ray irradiation were decreased by the Na and K co-doping. On the other hand, there was a small change within 940–1020 ns in the decay times by the Na and K co-doping. In the light yield proportionality under γ -ray irradiation, the non-proportionality in the low energy region was improved by Na and K co-doping.

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1. Introduction

Eu doped Srl₂ [Eu:Srl₂] scintillator crystal has high light yield (~80,000 ph/MeV) and high energy resolution (~3%) due to the small band-gap. Therefore, it is expected to be applied for next-generation radiation detector (Cherepy et al., 2008; Belli et al., 2012; Glodo et al., 2010; Cherepy et al., 2009; Hawrami et al., 2013; Gahane et al., 2009). The light yield proportionality of Eu:Srl₂ single crystal has been measured in the previous reports

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(Cherepy et al., 2008) and the Eu:SrI₂ single crystal indicated excellent light yield proportionality compared to the other scintillators as represented by Tl:NaI, Ce:LaBr₃ and Ce:Y₃Al₅O₁₂. In our previous report [Yokota et al., 2013, Nishimoto et al., 2014], the light yield proportionality of La co-doped Eu:SrI₂ crystals was measured and the non-proportionality became worse especially in the low energy region (<200 keV) by the La co-doping. On the other hand, there is a previous report about the effects of alkali metal (Na⁺) or alkali earth metal (Mg²⁺, Ca²⁺, Sr²⁺ and Ba²⁺) doping on the non-proportionality of the CeBr₃ scintillator crystal and the non-proportionality in the low energy region was improved by the doping (Quarati et al., 2014). According to the previous report, the improvement of the non-proportionality by the alkali metal or

alkali earth metal doping is originated from a more efficient e-h pair creation and/or energy transfer from the e-h pairs to the luminescent centers at energy below 100 keV. However, there is no report about the effects of alkali metal (Na⁺ and K⁺) co-doping to Eu:Srl₂ crystals on the optical and scintillation properties. According to the result of CeBr₃ crystal, the improvement of the non-proportionality is also expected for the Eu:Srl₂ crystal by the alkali metal.

On these backgrounds, in this study, we grew Na and K codoping Eu:SrI₂ crystals with various Na and K concentrations and measure the optical and scintillation properties to reveal the effects of the alkali metal co-doping on the optical and scintillation properties and improve the non-proportionality of Eu:SrI₂.

2. Experimental procedure

We grew the $\{(Sr_{1-x}A_x)_{0.925}Eu_{0.075}\}I_2$ [A,Eu:SrI $_2$, A = Na, K] crystals by a modified micro-pulling-down [μ -PD] method (Totsuka et al., 2012; Kral et al., 2013). Detail of the crystal growth procedure was described in the previous reports (Yokota et al., 2011). Mixed powders of the $\{(Sr_{1-x}A_x)_{0.925}Eu_{0.075}\}I_2$ with x = 0.00, 0.005, 0.01, and 0.05 were prepared from the starting materials of SrI $_2$ (4N), EuI $_2$ (5N), NaI (4N), and KI (4N) in the glove box filled with Ar (oxygen and moisture concentrations <1 ppm). By the modified μ -PD method, we grew the Non-doped, Na,Eu:SrI $_2$ and K,Eu:SrI $_2$ fiber crystals in high-purity Ar gas (99.9999%) at ~0.05 mm/min growth rate using a Pt—Rh wire as a seed without rotation. Polished specimens with 1 mm thickness were prepared from the first half parts of the grown fiber crystals by cutting and polishing in a mineral oil for measurements of optical and scintillation properties.

Parts of the grown crystals were ground in the globe box for X-ray diffraction [XRD] measurement by using an X-ray diffractometer (BRUKER, D8 DISCOVER) and a sealed sample chamber with no diffraction peak in the 2θ range of $20-70^\circ$. The configuration of the sealed sample chamber was a 51.5 mm outer diameter half-sphere made of polystyrol and polymethyl methacrylate. The X-ray source was $Cu-K_\alpha$ with accelerating voltage of 40 kV and beam current of 40 mA). The Inductively Coupled Plasma [ICP] analysis (Thermo Fisher Scientific, IRIS Advantage DIO) of the as-grown crystals was performed to identify the actual chemical compositions.

Room temperature emission and excitation spectra of the polished specimens were measured by using a spectrophotometers A10094 and Hamamatsu KK (Kurosawa et al., 2014). Pulse-height spectra and the decay curves of the specimens under γ -ray excitation from ¹³⁷Cs radiation source were measured to estimate their light yields and decay times using a photomultiplier [PMT](HA-MAMATSU, R7600U-200), multi-channel analyzer (AMPTEK CO. Pocket MCA 8000A) and an oscilloscope (TEKTRONIX, TBS1102). The measurements were performed in the grove box filled with Ar atmosphere, where the specimens and PMT were set. The shaping time was 2 µs in the pulse-height spectra measurements. The Ce 1 mol% doped Gd₂SiO₅ [Ce:GSO] single crystal with the light yield of 11,000 ph/MeV was used as a reference to estimate the light yields of the specimens. To evaluate the non-proportionality of the specimens, pulse-height spectra were measured with various γ -ray excitation sources (²²Na, ⁵⁴Mn, ⁶⁰Co, ¹³³Ba, ¹³⁷Cs, and ¹⁵²Eu). The non-proportionality of the Tl:NaI was also measured for the reference.

3. Results and discussions

Non-codoped Eu:SrI₂ and Na,Eu:SrI₂ and K,Eu:SrI₂ crystals with various Na and K concentrations were grown by the modified μ -PD method. It is shown in Fig. 1(a) that the as-grown and polished crystals. All as-grown crystals were approximately ϕ 2 mm in

diameter and more than 15 mm in length. No visible cracks and inclusions were observed in the polished non-codoped Eu:SrI₂, Na0.5% and 1.0%,Eu:SrI₂, and K0.5% and 1.0%,Eu:SrI₂ crystals and these crystals showing high transparency. On the other hand, milky parts in the polished Na5.0%, Eu: Srl₂ and K5.0%, Eu: Srl₂ crystals were present. It is shown in Fig. 1(b) that the powder XRD patterns of the as-grown crystals were measured for the phase identification. All diffraction peaks in the XRD patterns of the all crystals except for K5.0%,Eu:SrI₂ and Na5.0%,Eu:SrI₂ were identified by the SrI₂ crystal structure (Orthorhombic, Pnma). On the other hand, the additional diffraction peaks of secondary phases were observed in the XRD pattern of the K5.0%,Eu:SrI2 and Na5.0%,Eu:SrI2 crystals. The secondary phases of K5.0%, Eu: SrI₂ crystal were considered to be KSr₂I₅ and KI while the secondary phase(s) in the Na5.0%,Eu:SrI₂ crystal was not identified. According to the ICP analysis, the actual Na and K concentrations of Na5.0%, K1.0% and K5.0%, Eu:Srl₂ crystals were 1.3%, 0.3% and 2.7%, respectively. The actual Na and K concentrations of Na0.5%, Na1.0% and K0.5%, Eu: SrI₂ crystals were the below the limit of detection as it is shown in Table 1. In the Na, Eu: SrI2 and K,Eu:Srl₂ crystals, the actual Na and K concentrations were lower than the nominal composition due to the difference of the valences and ionic radius between Sr²⁺ and, Na⁺ and K⁺ ions. Therefore, another co-doping of the trivalence cation as represented by Y and La could be effective to improve the segregation of Na⁺ and K⁺.

We measured the room temperature photoluminescence spectra of the polished specimens (Fig. 2). In the emission spectra performed 410 nm excitation (Fig. 2(a)), an emission peak around 435 nm originated from 5d to 4f transitions of Eu²⁺ ion was observed for all crystals except for K5.0%,Eu:Srl₂. The emission peak for K5.0%,Eu:Srl₂ crystal was shifted to higher wavelength by the K co-doping. However, the reason is not clear now. In the excitation spectra monitored at 435 nm, the excitation peak was observed around 422 nm for all crystals. In addition, a broad excitation peak appeared around 315 nm for K5.0%,Eu:Srl₂ crystal. Probably originating from the secondary phase of Eu:KI (Aceves et al., 2005).

It is shown in Fig. 3 that the pulse-height spectra and decay curves of the non-codoped Eu:SrI₂, Na,Eu:SrI₂ and K,Eu:SrI₂ specimens under γ -ray irradiation. In the pulse-height spectra, clear photo-peaks were observed for the non-codoped, Na0.5%, Na1.0%, K0.5%, and K1.0%, Eu: Srl₂ specimens whereas there was no photopeak in the Na5.0% and K5.0%, Eu: Srl₂ crystals. The light yields of the specimens were estimated by using the Ce:GSO crystal (11,000 ph/MeV) as a reference (Fig.3(a)). The estimated light yields of the non-codoped, K0.5%, K1.0%, Eu: SrI₂ specimens and Na0.5%, Na1.0%,Eu:SrI₂ specimens were 70,000, 46,000, 48,500, 53,500 and 49,000 ph/MeV, respectively, and the light yield of the Eu:SrI₂ was decreasing by the Na and K co-doping. In addition, the estimated energy resolutions of the non-codoped, K0.5%, K1.0%, Eu: Srl₂ specimens and Na0.5%, Na1.0%, Eu: SrI₂ specimens were 11.3, 34.8, 11.6, 20.0 and 39.1%, respectively. The energy resolutions were relatively worse than previous reports about Eu:SrI₂. The results are considered to be due to the degradation of the surface for the polished specimens. The scintillation decay curves were fitted by the single exponential decay function (Fig. 3(b)) and the decay times were estimated. The decay times of all crystals were ranging from 940 to 1000 ns and there were few effects on the decay time with the Na and K co-doping as it was observed for the light yield evaluation as it is shown in Table 1.

The light yield proportionalities of the non-codoped Eu:Srl₂ and Na1.0%,Eu:Srl₂ and K1.0%,Eu:Srl₂ specimens were measured using various radiation sources emitting γ -ray as it is depicted in Fig. 4. Comparing to the result of non-codoped Eu:Srl₂, the non-proportionalities of Na1.0%,Eu:Srl₂ and K1.0%,Eu:Srl₂ improved in the low energy region (<200 keV). Especially, the effect of the improvement on the non-proportionality in the low energy region

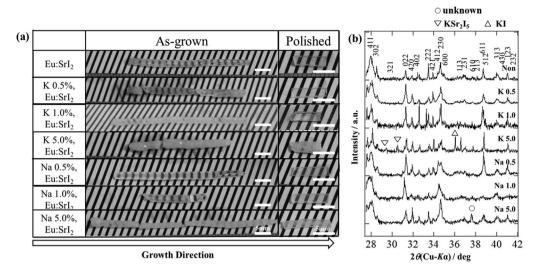


Fig. 1. (a) Non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ crystals grown by the μ-PD method. (b) Powder XRD patterns of the non-codoped grown by the μ-PD method. (c) Powder XRD patterns of the non-codoped grown by the μ-PD method grown by the

Table 1Summary of chemical composition, and optical and scintillation properties for the Na,Eu:Srl₂ and K,Eu:Srl₂ crystals.

Co- dopant	Nominal concentration of co- dopant/mo%	Actual concentration of co-dopant/mol% (measured by ICP)	Emission wavelength/nm	Excitation wavelength/nm	Light yield/ photons ⁺ MeV ⁻¹	Energy resolution/%	Decay time/ns
K	0	<0.1	435	420	70,000	11.3	959
	0.5	<0.1	435	422	46,000	34.8	947
	1.0	0.3	436	421	48,500	11.6	941
	5.0	2.7	435	310,424	_	_	968
Na	0.5	<0.1	435	421	53,500	20.0	968
	1.0	<0.1	435	422	49,000	39.1	1020
	5.0	1.3	435	424	_	_	947

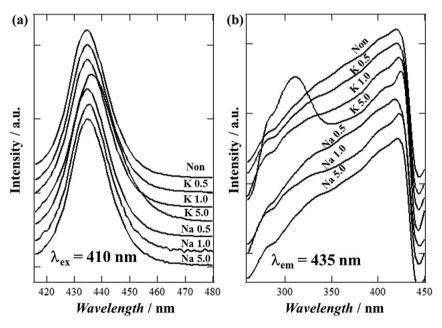


Fig. 2. (a) Emission and (b) excitation spectra of the non-codoped, Na,Eu:SrI2 and K,Eu:SrI2 specimens.

by the K co-doping was larger than that of the Na co-doping. The results suggest that alkali metal co-doping made the e-h pair cr1eation and/or energy transfer from the e-h pairs to the luminescent centers at energy below 100 keV more efficient and the effect is dependent on the ionic radius of co-dopant.

4. Conclusions

We grew the Na and K co-doped Eu:SrI $_2$ crystals by the modified μ -PD method and their optical and scintillation properties were measured. The grown non-codoped Eu:SrI $_2$, Na,Eu:SrI $_2$ and

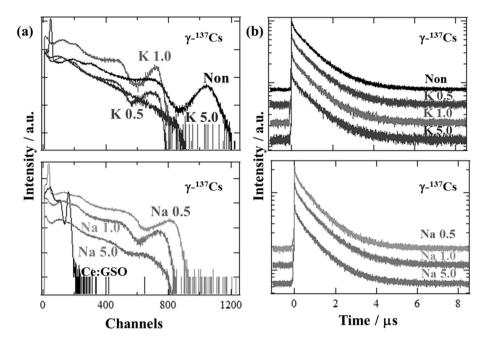


Fig. 3. (a) Pulse-height spectra and (b) decay curves of the non-codoped, Na,Eu:Srl₂ and K,Eu:Srl₂ specimens under γ-ray irradiation.

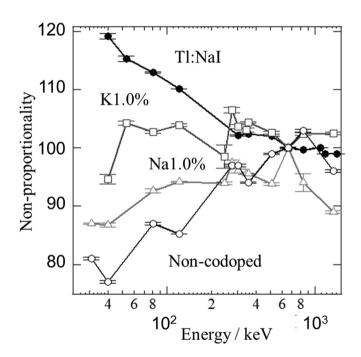


Fig. 4. Light yield proportionality of the non-codoped, Na,Eu:Srl $_2$ and K,Eu:Srl $_2$ specimens under γ -ray irradiation.

K,Eu:SrI₂ crystals except for the Na5.0% and K5.0%,Eu:SrI₂. Grown crystals indicated high transparency while the milky parts were generated in the Na5.0% and K5.0%,Eu:SrI₂ crystals as indicated by presence of secondary phases in the powder XRD patterns. In the photoluminescence spectra, emission and excitation peaks were observed around 435 nm and 422 nm respectively for all specimens except for K5.0%,Eu:SrI₂. On the other hand, K5.0%,Eu:SrI₂ showed the emission peak at the longer wavelength and the additional excitation peak around 330 nm which was considered to be attributable to the secondary phases. In the pulse-height spectra, the light yield of the Eu:SrI₂ was decreased by the Na and K co-

doping whereas the decay times were almost constant (940–1020 ns). Meanwhile, the non-proportionality of the Eu:Srl $_2$ was improved in the low energy region (<200 keV) by the Na and K co-doping and the K1.0%,Eu:Srl $_2$ crystal showed the most excellent non-proportionality.

Acknowledgments

This work is partially supported by the New Energy and Industrial Technology Development Organization (NEDO), the Japan Society for the Promotion of Science (JSPS) Research Fellowships for Grant-in-Aid for Young Scientists, Bilateral AS CR-JSPS Joint Research Project, MEYS, KONTAKT II, no. LH14266, Japan Science and Technology Agency (JST) Adaptable & Seamless Technology Transfer Program through Target-driven R&D (A-STEP). In addition, we would like to thank following person for the support: Mr. Tetsuya Ashio in Analytical Research Core for Advanced Materials, Tohoku University.

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